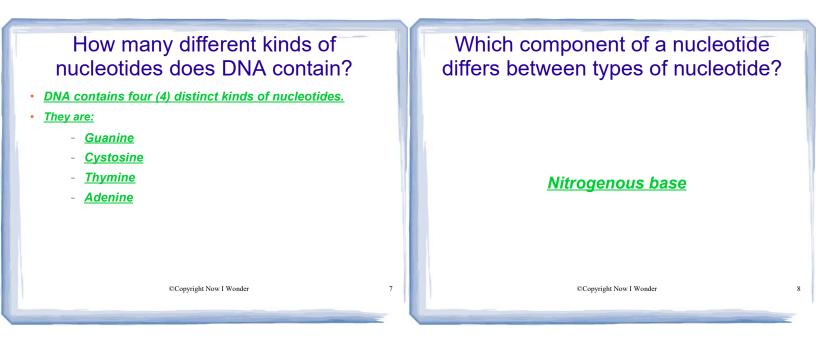


What is a nucleotide composed of? (Select all that apply)

- Phosphate Cholesterol
- <u>Nitrogenous</u>
 ATP

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Describe the different kinds of nucleotides found in DNA.

The four (4) kinds of nucleotide found in DNA vary by size and by the chemical make-up of their nitrogenous bases.

The two larger nucleotides are adenine and guanine. They are double-ringed structures called "purines".

<u>The two smaller nucleotides are cytosine and thymine.</u> <u>They are smaller because they contain only single-ring</u> <u>structures and are called "pyrimidines".</u>

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How does a nucleic acid form?

<u>Nucleic acids are long polymers (large</u> <u>molecules formed of long chains of similar</u> <u>molecules called subunits) of repeating</u> <u>subunits called nucleotides.</u>

<u>Nucleic acids are formed when the individual</u> <u>sugars link together in a long line by their</u> <u>phosphate group (the long line is called a</u> <u>polynucleotide chain).</u>

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How does the structure of a nucleic acid allow it to store information?

•Nucleic acids act like books.

Individual nucleotides act as letters of the alphabet.

•<u>The pattern and position of different nucleotides in each</u> polynucleotide chain act as words and code for the creation of <u>different proteins.</u>

•<u>Nucleic acids encode information by varying which nucleotide is</u> positioned at each point in the polymer.

•<u>Strung together, the pattern and order of the individual</u> <u>nucleotides can be "read" to provide a huge amount of</u> <u>information, just like a novel.</u>

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Which of the following best describes the structure of DNA

- 2 polynucleotide chains composed of mononucleotides <u>covalently bonded between the sugar of one and the</u> <u>phosphate of another mononucleotides to form a sugar-</u> <u>phosphate backbone. Complementary nitrogenous bases</u> <u>attached to the sugar-phosphate backbone point inward</u> <u>toward each other and are linked by hydrogen bonds.</u>
- 2 nucleotides bonded to the sugar molecule of a carbon isotope by covalent bonds. The isotopes form a sugar-carbon backbone. Complementary phosphoric bases attached to the sugar-carbon backbone point inward toward each other and are linked by covalent bonds.

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